

Atom-Efficient Assembly of 1,5-Oxygen-Bridged Medium-Sized Carbocycles by Sequential Combination of a Ru-Catalyzed Alkyne–Alkene Coupling and a *Prins*-Type Cyclization

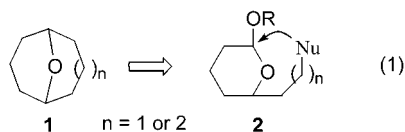
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Received December 18, 2001

Medium-sized carbocycles, particularly eight- and nine-membered ones, form the structural cores of numerous biologically relevant natural products and are therefore important synthetic targets.¹ The well-known difficulty of their assembly by conventional cyclization routes, which is due to unfavorable entropic and enthalpic factors,² makes the development of new, practical approaches to these sized rings a major synthetic challenge.

We envisaged that the energy cost of the cyclization step might be considerably decreased if it is templated by the introduction of a temporary internal tether in the acyclic precursor. Thereby, a *Prins*-type cyclization of a mixed acetal such as **2** could lead to these sized carbocycles (**1**) because in effect the cyclization proceeds to form a six- or seven-membered oxacycle (eq 1). The oxygen



bridge of **1** was thought particularly interesting because it imposes a high degree of conformational rigidity that might facilitate stereoselective manipulations at the carbocycle prior to untethering.³

Herein we demonstrate that a ruthenium-catalyzed coupling of 1-trimethylsilyl-1-alkyn-3-ols to commercially available allyl ethyl ether provides a rapid, atom-economical⁴ and versatile method for preparing the required cyclization precursors. Lewis acid-induced closure of the resulting products leads to the expected oxabicycles, compounds that can be easily manipulated to unmask the underlying eight- and nine-membered carbocycles.

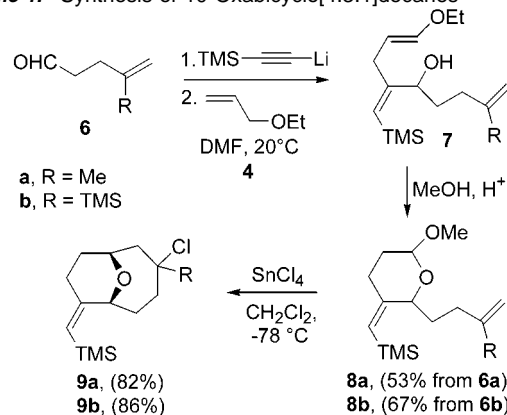
A key step in the genesis of the new procedure was the realization that 5-hydroxy-1-enol ethers susceptible to straightforward conversion to mixed acetals of type **2** might be obtained by using 1-trimethylsilyl-1-alkyn-3-ols and allyl ethers as reacting partners in the recently developed ruthenium-catalyzed alkyne–alkene coupling method.⁵ However, reaction of alkyne **3** and allyl ethyl ether (1:1) with 10 mol % of catalyst **4**, in acetone (0.1 M) at room temperature,^{5a} gave a quite complex mixture of products that included no more than traces of the desired alcohol **5**. In 5:1 THF/acetone the reaction proceeded with low conversion to give a 15% yield of **5**, with mostly unreacted starting material (Table 1, entry 2). Other alkenes such as allyl acetate or allyl alcohol gave mixtures of unwanted products in addition to unreacted starting material. We later found that using DMF as solvent the reaction is much more efficient,⁶ and even more so when higher concentrations of the reactants were used (entry 4). Best results were obtained with

Table 1. Ruthenium-Catalyzed Coupling of **3** with Allyl Ethyl Ether^a

entry	solvent	concn of 3 (M)	allyl ether (equiv)	4 (%)	ratio 5:3 ^b	yield ^c (%)
1	acetone	0.1	1	10		
2	THF/acetone (10:1)	0.1	1	10	1:4	15 (50) ^d
3	DMF	0.2	1	10	1.6:1	48
4	DMF	1	1	10	2.3:1	66 (91)
5	DMF	1	1	1		8
6	DMF	1	1	5	1.5:1	48 (72)
7	DMF/H ₂ O (10:1)	1	1	10	2.3:1	60 (85)
8	DMF	1	1.5	10	2.7:1	73 (98)
9	DMF	1	3	10	1:1	49 (97)

^a All reactions were carried out by adding catalyst **4** to a solution of **3** and allyl ethyl ether, and were stopped after 2 h at room temperature. ^b Calculated by ¹H NMR after workup. ^c Isolated yield. ^d Yield based on chromatographically recovered alkyne.

Scheme 1. Synthesis of 10-Oxabicyclo[4.3.1]decanes



1.5 equiv of the allyl ether partner (entry 8). We have also observed that the order of addition is important (Table 1, footnote a): adding the allyl ether to a mixture of the catalyst and the alkyne in DMF gave a significantly lower yield (about 30%).⁷

Once the alkyne–alkene coupling had been optimized, its application to alkyne–alkene coupling with a nucleophilic alkene suitable for carrying out the planned *Prins* cyclization⁸ was investigated. Gratifyingly, the enol resulting from the addition of trimethylsilyllithium acetylide to aldehyde **6a** coupled efficiently with allyl ethyl ether, giving the expected enol ether **7a**. This product was

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